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(54) **Extrudable thermoplastic composition consisting of a compatibilized polyphenylene ether polyamide resin blend.**

(57) The present invention deals with an extrudable thermoplastic composition consisting of (A) a compatibilized polyphenylene etherpolyamide resin blend. By using a polyphenylene ether with an intrinsic viscosity of more than 44ml/g as measured in toluene at 25 degrees C and a concentration of 0.6 gram per 100 ml and a polyamide with a reduced viscosity of more than 175 ml/g as measured in sulphuric acid in accordance with ISO 307, (B) 0-45% by weight of reinforcing fibers and (C) up to 2% by weight of one or more of the following additives: stabilizers, dyes, pigments, release agents, wherein all quantities have been calculated with respect to the sum of A, B and C, with the exception of compositions comprising an epoxy compound.

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Th invention relates to thermoplastic compositions consisting of (A) a compatibilized polyphenyl n-ether-polyamide resin blend and; (B) 0-45% by weight of reinforcing fibers and (C) up to 2% by weight of one or more of the following additives: stabilizers, dyes, pigments, release agents, where all quantities have been calculated with respect to the sum of A, B, and C, with the exception of compositions comprising an epoxy compound.

The invention also relates to articles formed out of the compositions of the invention.

Commercially available blends of polyphenylene ether and polyamide are less suitable for molding into the desired shapes by extrusion; they are manufactured by a compounding-extrusion step, followed by pelletization. The obtained pellets are then usually processed by injection molding into the desired shapes.

The present invention deals with compositions which can be formed by extrusion into the desired shape.

It is generally known to increase the molecular weight and hence the viscosity of thermoplastic resins to make them better suitable for molding by extrusion. The use of thermoplastics resins with increased viscosity results in higher melt strength values as necessary for extrusion-molding.

The commercially most interesting blends of polyphenylene ether and polyamide comprise a continuous phase of polyamide in which the polyphenylene ether is dispersed. It would thus be expected that an increase of the viscosity of the polyamide only would make the blend more suitable for molding by extrusion. An increase of the melt viscosity did indeed occur by using a polyamide with increased viscosity in combination with a polyphenylene ether of standard viscosity. Unfortunately this blend with a normal polyphenylene ether showed a very undesirable effect: upon extrusion in the manufacturing step so-called die-swell occurred to such an extent that the product was not uniform in thickness and not well processable.

EP-A-0 373 900 describes a composition comprising a polyphenylene ether, a polyamide, a rubber-like material and an organic compound containing an oxirane oxygen atom. The organic compound is added to improve the moldability by extrusion low molding.

The intrinsic viscosity of the polyphenylene ether may vary between 0,35 to 0,75 dl/g, whereas the relative viscosity of the polyamide should be at least 2,5. An epoxy compound can be used as the oxirane oxygen atom containing compound.

In the comparative example 7 has been shown that the absence of the compound with an oxirane oxygen atom results in breakage of a parison in blow molding applications. This shows that the known composition has insufficient melt strength thus making it unsuitable for extrusion purposes.

It has now been found that the composition of the invention which consists of the components A, B and C and thus does not comprise any further constituents such as a rubber like material described in EP-A-373 900 is, even in the absence of oxirane oxygen atom containing compound, suitable for the shaping of articles by means of extrusion.

It has now been found that by using a combination of a polyphenylene ether with a specific viscosity which is higher than usual and of a polyamide with a viscosity which is higher than usual together with 0-45% by weight of glass fibers and up to 2% weight of additives it is possible to obtain a blend with a satisfactory melt strength without occurrence of the die-swell phenomena.

In the thermoplastic composition of the invention the polyphenylene ether has an intrinsic viscosity of more than 44 ml/g as measured in toluene at 25 degrees C and a concentration of 0.6 gram per 100 ml and the polyamide has a reduced viscosity of more than 175 ml/g as measured in sulphuric acid in accordance with ISO 307.

The polyphenylene ether in the blends of the invention preferably have an intrinsic viscosity of at least 50 ml/g. The polyamide preferably has a reduced viscosity of at least 190 ml/g.

EP-A-270 796 deals with composition comprising a polyphenylene ether and a polyamide. To increase the impact strength a mixture of two types of polyamides has been used.

In the examples a high viscosity polyamide and high viscosity polyphenylene ether has been used in combination with other ingredients like a rubbery polymer. The presence of a rubbery polymer makes the compositions of the examples less suitable for molding by extrusion.

Preferred polyphenylene ether resins (PPE) and polyamide resins as well as means for providing compatibilized combinations thereof are described below.

In general it is desirable that the polyamide component forms the continuous phase in the blend and, therefore, typically at least 35 percent by weight of the total PPE-polyamide- composition will be comprised of the polyamide component.

Polyphenyl ethers are a well known class of compounds sometimes referred to as polyphenylene oxide s. Examples of suitable polyphenylene ethers and processes for their preparation can be found in US A 3,306,874; 3,306,875; 3,257,357; and 3,257,358. Compositions of the present invention will encompass homopolymers, copolymers and graft copolymers obtained by the oxidative coupling of phenolic com-

pounds. The preferred polyphenyl ether resins used as base resins in compositions of the present invention will be comprised of ml/g derived from 2,6-dimethyl phenol. Also contemplated are polyphenyl ether copolymers comprising of ml/g derived from 2,6-dimethyl phenol and 2,3,6-trimethyl phenol.

A particularly useful polyphenylene ether would be poly(2,6-dimethyl-1,4-phenylene ether) having an intrinsic viscosity (I.V.) of more than 50 ml/g as measured in toluene at 25 degrees C and a concentration of 0.6 gram per 100 ml.

The polyamide resins useful in the practice of the present invention are a generic family of resins known as nylons, characterized by the presence of an amide group (-CONH-). Nylon-6 and nylon-6,6 are the generally preferred polyamides and are available from a variety of commercial sources. Other polyamides, however, such as nylon-4, nylon-12, nylon-6,10, nylon 6,9 or others such as the amorphous nylons may be useful for particular polyphenylene ether-polyamide applications. The most preferred polyamide for the blends of the present invention is the polyamide-6,6.

The polyamides can be obtained by a number of well known processes. Nylon-6, for example, is a polymerization product of caprolactam. Nylon-6,6 is a condensation product of adipic acid and hexamethylenediamine. A nylon-6,6 having reduced viscosity of more than 175 ml/g, preferably of at least 190 ml/g measured in sulphuric acid in accordance with ISO 307 is the most preferred type of polyamide.

In blends of the present invention, a compatibilizing agent should be employed in the preparation of the composition. The two-fold purpose for using compatibilizing agents is to improve, in general, the physical properties of the polyphenylene ether-polyamide resin blend, as well as to enable the use of a greater proportion of the polyamide component. When used herein, the expression "compatibilizing agent" refers to those polyfunctional compounds which interact with either the polyphenylene ether, the polyamide, or both. This interaction may be chemical (e.g. grafting) or physical (e.g. affecting the surface characteristics of the dispersed phases). In either instance the resulting polyphenylene ether-polyamide composition appears to exhibit improved compatibility, particularly as evidenced by enhanced impact strength, mold knit line strength and/or elongation. As used herein, the expression "compatibilized polyphenylene ether-polyamide base resin" refers to those compositions which have been physically or chemically compatibilized with an agent as discussed above, as well as those compositions which are physically compatible without such agents, as taught in US-A-3,379,792.

Examples of the various compatibilizing agents that may be employed in the practice of the present invention include: a) liquid diene polymers, b) oxidized polyolefin wax, c) quinones, d) organosilane compounds, e) polyfunctional compounds and functionalized polyphenylene ethers obtained by reacting one or more of the previously mentioned compatibilizing agents with a polyphenylene ether as described hereinafter.

Liquid diene polymers (a) suitable for use herein include homopolymers of a conjugated diene with at least one monomer selected from the group consisting of other conjugated dienes; vinyl monomer, e.g. styrene and alphamethyl styrene; olefins, e.g. ethylene, propylene, butene-1, isobutylene, hexene-1, octene-1 and dodecene-1, and mixtures thereof, having a number average molecular weight of from 150 to 10,000, preferably 150 to 5,000. These homopolymers and copolymers can be produced by the methods described in, for example, US-A-4,054,612; 3,876,721 and 3,428,699 and include, among others, polybutadiene, polyisoprene, poly(1,3-pentadiene), poly(butadiene-isoprene), poly(styrene-butadiene), polychloroprene, poly(butadiene-alpha methylstyrene), poly(butadiene-styrene-isoprene), poly(butylene-butadiene) and the like.

Oxidized polyolefin waxes (b) are well known and a description thereof and processes for the production of the same are found in US-A-3,822,227 and 3,756,999 and German Patent Publications 3,047,915 and 2,201,862.

Generally, these are prepared by an oxidation or suspension oxidation of polyolefin. An especially preferred polyolefin wax is "Hoechst Wachs".

Quinone compounds (c) suitable for use herein are characterized as having in the molecule of the unsubstituted derivative at least one 6 membered carbon ring; at least two carbonyl groups in the ring structure, both of which may be in the same or, if more than one ring, different rings, provided that they occupy positions corresponding to the 1,2- or 1,4-orientation of the monocyclic quinone; and at least two carbon-carbon double bonds in the ring structure, said carbon-carbon double bonds and carbonyl carbon-carbon double bonds in the ring structure, said carbon-carbon double bonds and carbonyl carbon-oxygen double bonds being conjugated with respect to each other. Where more than one ring is present in the unsubstituted quinone, the rings may be fused, non-fused or both; non-fused rings may be bound by a direct carbon-carbon double bond or by a hydrocarbon radical having conjugated unsaturation such as =C-C=.

Substituted quinones are also within the scope of the present invention. The degree of substitution, where substitution is desired, may be from one to the maximum number of replaceable hydrogen atoms. Exemplary of the various substituents that may be present on the unsubstituted quinone structures include halogen, e.g. chlorin, bromin, fluorin, etc. hydrocarbon radicals including branched and unbranched, 5 saturated and unsaturated alkyl, aryl, alkyl aryl and cycloalkyl radicals and halogenated derivatives thereof; and similar hydrocarbons having hetero atoms therein, particularly oxygen, sulfur or phosphorous and wherein the same connects the radical to the quinone ring (e.g. oxygen link).

Exemplary of the various quinones there may be given 1,2- and 1,4-benzoquinone; 2,6-diphenyl quinone; tetramethyldiquinone; 2,2'- and 4,4'-diphenoxquinone; 1,2-, 1,4- and 2,6-naphthoquinone; chloranils; 10 2-chloro-1,4-benzoquinone; 2,6-dimethyl benzoquinone and the like.

Organosilane compounds (d) suitable as compatibilizing agents are characterized as having in the molecule (a) at least one silicon atom bonded to a carbon through an oxygen link and (b) at least one carbon-carbon double bond or carbon-carbon triple bond and/or a functional group selected from the group consisting of an amine group or a mercapto group provided that the functional group is not directly bonded 15 to the silicon atom.

In such compounds, the C-O-Si component is generally present as an alkoxy or acetoxy group bonded directly to the silicon atom, wherein the alkoxy or acetoxy group generally has less than 15 carbon atoms and may also contain hetero atoms (e.g. oxygen). Additionally, there may also be more than one silicon atom in the compound, such multiple silicon atoms, if present, being linked through an oxygen link (e.g. 20 siloxanes), a silicon bond; or a bifunctional organic radical (e.g. methylene or phenylene groups).

Examples of suitable organosilane compounds include: gamma amino propyltriethoxy silane, 2-(3-cyclohexanyl)ethyl trimethoxy silane; 1,3-divinyl tetraethoxy silane; vinyl tris-(2-methoxyethoxy)silane; 5-bicycloheptenyl triethoxy silane and gamma mercapto propyl trimethoxy silane.

Polyfunctional compounds (e) which may be employed as compatibilizer in the practice of the present 25 invention are of three types. The first type of polyfunctional compounds are those having in the molecule both (a) a carbon-carbon double bond or a carbon-carbon triple bond and (b) at least one carboxylic acid, acid anhydride, acid halide, anhydride, acid halide anhydride, acid amide, acid ester, imide, amino, or hydroxy group. Examples of such polyfunctional compounds include maleic acid; maleic anhydride; fumaric acid; citraconic acid; itaconic acid; maleimide; maleic hydrazide; reaction products resulting from a diamine 30 and maleic anhydride, maleic acid, fumaric acid, etc.; dichloro maleic anhydride; maleic acid amide; unsaturated dicarboxylic acids (e.g. acrylic acid, butenoic acid, methacrylic acid, t-ethylacrylic acid, pentenoic acid); decenoic acids, undecenoic acids, dodecenoic acids, linoleic acid, etc.); esters, acid amides or anhydrides of the foregoing unsaturated carboxylic acids; unsaturated alcohols (e.g. alkyl alcohol, crotyl alcohol, methyl vinyl carbinol, 4-pentene-1-ol, 1,4-hexadiene-3-ol, 3-butene-1,4-diol, 2,5-dimethyl-3-hexene-2,5-diol and alcohols of the formula  $C_nH_{2n-5}OH$ ,  $C_nH_{2n-7}OH$  and  $C_nH_{2n-9}OH$ , wherein n is a positive 35 integer up to 30), unsaturated amines resulting from replacing from replacing the -OH group(s) of the above unsaturated alcohols with NH<sub>2</sub> groups; and functionalized diene polymers and copolymers. Of these, one of the preferred compatibilizing agents for compositions of the present invention is maleic anhydride.

The second group of polyfunctional compatibilizer compounds suitable for use herein are characterized 40 as having both (a) a group represented by the formula (OR) wherein R is hydrogen or an alkyl, aryl, acyl or carbonyl dioxy group and (b) at least two groups each of which may be the same or different selected from carboxylic acid, acid halide, acid anhydride, anhydride, acid halide anhydride, acid ester, acid amide, imido, amino and salts thereof. Typical of this group of compatibilizers are the aliphatic polycarboxylic acids, acid esters and acid amides represented by the formula:

45  $(R^I O)_m R(COOR^{II})_n (CONR^{III} R^{IV})_s$

wherein R is a linear or branched chain, saturated aliphatic hydrocarbon of from 2 to 20, preferably 2 to 10, carbon atoms; R<sup>I</sup> is selected from the group consisting of hydrogen or an alkyl, aryl, acyl or carbonyl dioxy group of 1 to 10, preferably 1 to 6, most preferably 1 to 4, carbon atoms, especially preferred is hydrogen; each R<sup>II</sup> is independently selected from the group consisting of hydrogen or an alkyl or aryl group from 1 to 20 carbon atoms, preferably from 1 to 10 carbon atoms; each R<sup>III</sup> and R<sup>IV</sup> is independently selected from the group consisting essentially of hydrogen or an alkyl or aryl group of from 1 to 10, preferably from 1 to 6, most preferably 1 to 4, carbon atoms; m is equal to 1 and (n + s) is greater than or equal to 2, 50 preferably equal to 2 or 3, and n and s are each greater than or equal to zero and wherein (OR<sup>I</sup>) is alpha or beta to a carbonyl group and at least two carbonyl groups are separated by 2 to 6 carbon atoms. Obviously, R<sup>I</sup>, R<sup>II</sup>, R<sup>III</sup> and R<sup>IV</sup> cannot be aryl when the respective substituent has less than 6 carbon atoms.

Illustrative of suitable polycarboxylic acids there may be given citric acid, malic acid, and agaric acid; including the various commercial forms thereof, such as for example, the anhydrous and hydrated acids. Of these, citric acid is another of the preferred compatibilizing agents. Illustrative of acids useful herein include for example, acetyl citrate and mono- and/or diisopropyl citrates and the like. Suitable acid amides useful herein include for example N,N'-diethyl citric acid amide; N-phenyl citric acid amide; N-dodecyl citric acid amide; N,N'-didecyl citric acid amide and N-dodecyl malic acid present invention. Especially preferred derivatives are the salts thereof, including the salts with amines and/or preferably, the alkali and alkaline metal salts. Exemplary of suitable salts include calcium malate, calcium citrate, potassium malate and potassium citrate.

10 The third group of polyfunctional compatibilizer compounds suitable for use herein are characterized as having in the molecule both (a) an acid halide group, most preferably an acid chloride group and (b) at least one carboxylic acid, carboxylic acid anhydride, acid ester or acid amide group, preferably a carboxylic acid or carboxylic acid anhydride group. Exemplary of compatibilizers within this group there may be given trimellitic anhydride acid chloride, chloroformyl succinic anhydride, chloroformyl succinic acid, chloroformyl glutaric anhydride, chloroformyl glutaric acid, chloroacetyl succinic anhydride, chloroacetyl succinic acid, trimellitic acid chloride and chloroacetyl glutaric acid. Among these, trimellitic anhydride acid chloride is preferred. Furthermore, it is especially preferred that compatibilizers of this group be prereacted with at least a portion of the polyphenylene ether whereby the compatibilizing agent is a PPE-functionalized compound.

20 Each of the foregoing compatibilizing agents are more fully described in US-A-4,315,086 and 4,642,358; and European Patent Application No. 04640.

The foregoing compatibilizing agents may be used alone or in any combination of one another. Furthermore, they may be added directly to the melt blend or precompounded with either or both the polyphenylene ether and polyamide as well as with other resinous materials employed in the preparation of the compositions of the present invention. With many of the foregoing compatibilizing agents, particularly the polyfunctional compounds, even greater improvement in compatibility is found where at least a portion of the compatibilizing agent is precompounded with all or part of the polyphenylene ether. It is believed that such precompounding may cause the compatibilizing agent to react with the polymer and, consequently, functionalize that polymer as noted above for example, the polyphenylene oxide may be precompounded with trimellitic anhydride acid chloride to form an anhydride functionalized polyphenylene ether which has improved compatibility with the polyamide compared to a non-functionalized ether.

30 Where the compatibilizing agent is employed in the preparation of the compositions of the present invention, the initial amount used will be dependent upon the specific compatibilizing agent chosen and the specific polymeric system to which is added.

35 It is possible to use in the composition according to the invention any other known compatibilisation system, with the exception of epoxy compounds. Other systems have been described for example in US-4,866,114.

The thermoplastic composition of the invention may comprise any of the following additives: reinforcing fibers, such as glass fibers.

40 Glass reinforced grades of compatibilized polyphenylene etherpolyamide resin compositions are extremely important in thermoplastic applications requiring a unique combination of physical properties such as impact strength, rigidity and modulus, dimensional stability, high heat resistance as well as chemical resistance. Principally, glass filled PPE-polyamide compositions of the present invention offer good physical properties. Ten to thirty weight percent loadings of chopped glass fiber can typically be utilized to advantage in these compositions. These weight percents would be based upon the weight of the polyphenylene ether resin, the polyamide resin, and the glass, taken together. More or less glass (e.g. 5 to 45 weight percent) can also be utilized in certain circumstances. Less than about 5 parts glass begins to perform like an unreinforced product. More than about 45 weight percent glass begins to interfere with the ability of the resin to coat and bind the glass effectively.

50 The composition of the invention may further comprise up to 2% by weight of one or more of the following additives, such as stabilizers, dyes, pigments, release agents.

All patents and patent applications mentioned above are incorporated herein by reference.

The invention will be further illustrated by the following examples.

## 55 Examples

In the examples the following materials have been used:

PPE-1: a poly(2,6-dimethyl-1,4-phenylene ether) with an intrinsic viscosity of 40 ml/g as measured in

toluen at 25 degrees C and a concentration of 0.6 gram per 100 ml PPO 803);  
 PPE-2: a poly(2,6-dimethyl-1,4-phenylene ether) with an intrinsic viscosity of 45 ml/g as measured in toluene at 25 degrees C and a concentration of 0.6 gram per 100 ml PPO 800);  
 PPE-3: a poly(2,6-dimethyl-1,4-phenylene ether) with an intrinsic viscosity of 57 ml/g as measured in toluene at 25 degrees C and a concentration of 0.6 gram per 100 ml PPO 805);  
 PA-1: a polyamide-6,6 with a reduced viscosity of 145 ml/g as measured in sulphuric acid in accordance with ISO 307.(A3Q44)  
 PA-2: a polyamide-6,6 with a reduced viscosity of 205ml/g as measured in sulphuric acid in accordance with ISO 307.(A4)  
 PA-3: a polyamide-6,6 with a reduced viscosity of 270ml/g as measured in sulphuric acid in accordance with ISO 307.(A5)

CA: citric acid.

ST: standard iodine based stabilizers

PETS: penta-erythritol tetrastearate

CB: carbon black

The ingredients were compounded in the quantities as indicated in the following table in a twin screw extruder with six different temperature settings over the length of the extruder, varying between 280 and 310 degrees C. The screw speed was 300 rounds per minute, the throughput 10 kilograms per hour. All ingredients with exception of the polyamide were feeded at the throat of the extruder; the polyamide was feeded downstream about halfway the extruder.

The die of the extruder had a diameter of 4 mm. Some of the blends coming out of the extruder suffered from die-swell, meaning that the strands had a varying diameter far above 4 mm. This is an entirely undesirable effect, since this causes great instability. This makes the blends unsuitable for molding formed products out of them by means of extrusion and unsuitable for pelletizing them in a regular production process.

The strands coming from the extruder were pelletized in a laboratory equipment and dried for 3 hours at 110 degrees C.

The dried pellets were brought into the extruder of a Rheotens melt viscosity instrument of Goettfert-Feinwerk-Technik GmbH. The apparatus comprises an extruder with an extrusion head with a die diameter of 4 mm and two wheels placed at a distance of 120 mm below the extrusion head. The strand is conducted between the wheels which turn around with increasing speed. The tensile speed and the tensile force are monitored. Upon occurrence of an instability they are measured; the melt strength in Newton is calculated based on the measured values. The results are also given in the table here below.

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TABLE

Example	A*	B*	C*	I	II	III	IV
Composition (Parts by weight)							
PPE-1:	58.9	58.9	-	-	-	-	-
PPE-2:	-	-	-	-	-	-	58.9
PPE-3:	-	-	58.9	58.9	58.9	53.9	-
PA-1:	40	-	40	-	-	-	-
PA-2:	-	40	-	40	-	45	40
PA-3:	-	-	-	-	40	-	-
CA :	0.7	0.7	0.7	0.7	0.7	0.7	0.7
ST :	0.1	0.1	0.1	0.1	0.1	0.3	0.1
PETS:	0.3	0.3	0.3	0.3	0.3	0.3	0.3
CB :	0.1	0.1	0.1	0.1	0.1	0.1	0.1
Properties							
Die Swell	no n.d.	yes 26	no 22	no 26	no 35	no n.d.	minor n.d.
Melt Strength							

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\* n.d. means not determined. Examples A, B and C are comparative examples.

As can be seen from the results of the table the common blends of polyphenylene ether and a polyamide (Example A) can be readily extruded from the compounding machine. Their melt strength is

how v r unsufficient to shape articles out of this known blend by xtrusion.

The bl nd of Example B with a standard polyph nylene th r and a high viscosity polymid does show a good m lt str ngth, as could be xp ct d sinc th polyamide forms th continuous phase. Quite unexpectedly th comparativ blend showed strong di -sw ll making it unusabl for xtrusion purposes.

The above results are quite surprising since the polyamide with high viscosity as used in comparative example B itself i.e. without being blended with a polyphenylene ether is perfectly suitable for extrusion purposes.

Using a polyphenylene ether with higher viscosity in combination with a standard polyamide (blend C) results in a blend without die-swell, but with a rather low melt strength.

Only by combining a high viscosity polyphenylene ether with a high viscosity polyamide (Examples I, II, Illand IV) was it possible to obtain a material with a good melt strength, without die swell or with minor die-swell only..

### Claims

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1. An extrudable thermoplastic composition consisting of (A) a compatibilized polyphenylene ether-polyamide resin blend, wherein the polyphenylene ether has an intrinsic viscosity of more than 44 ml/g as measured in toluene at 25 degrees C and a concentration of 0.6 gram per 100 ml and wherein the polyamide has a reduced viscosity of more than 175 ml/g as measured in sulphuric acid in accordance with ISO 307; (B) 0-45% by weight of reinforcing fibers and (C) up to 2% by weight of one or more of the following additives: stabilizers, dyes, pigments, release agents, wherein all quantities have been calculated with respect to the sum of A, B and C, with the exception of compositions comprising an epoxy compound.

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2. A composition as in claim 1 wherein the polyphenylene ether has an intrinsic viscosity of at least 50 ml/g and the polyamide has a reduced viscosity of at least 190 ml/g.

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3. A composition as in Claim 1 wherein the compatibilized resin blend consists of 5 to 95 weight percent polyphenylene ether resin and 95 to 5 weight percent polyamide resin, based upon the weight of polyphenylene ether and polyamide together.

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4. A composition as in Claim 3 wherein said polyamide resin constitutes a continuous phase in an amount greater than 35 weight percent of the compatibilized resin blend.

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5. A composition as in Claim 1 comprising a compatibilizing agent selected from the group consisting of: maleic anhydride, fumaric acid, citric acid, malic acid, and reaction products of a polyphenylene ether and trimellitic anhydride acid chloride.

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6. A composition as in Claim 1 wherein said polyphenylene ether resin is a polymer or copolymer comprised of one or more units derived from units selected from the group consisting of 2,6-dimethyl phenol units and 2,3,6-trimethyl phenol units.

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7. A composition as in claim 6 wherein said polyphenylene ether is poly(2,6-dimethyl-1,4-phenylene ether).

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8. A composition as in Claim 1 wherein said polyamide resin is a polyamide 6,6.

9. Articles formed out of the composition of claim 1.



European Patent  
Office

## EUROPEAN SEARCH REPORT

Application Number  
EP 95 20 0729

DOCUMENTS CONSIDERED TO BE RELEVANT					
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)		
X	EP-A-0 416 436 (BASF AG) * page 3, line 15 - line 20 * * page 3, line 29 - line 31 * * page 4, line 10; claim 1 * ---	1-9	C08L71/12 C08L77/00 C08K3/00 C08K5/00 C08K7/02		
X	EP-A-0 501 175 (BASF AG.) * claim 1 * * column 3, line 50 - column 4, line 5 * * table 1 * * column 4, line 20 - line 25 * * column 14, line 8 - line 20 * ---	1-9			
A	EP-A-0 454 156 (MITSUBISHI PETROLCHEM. CO.) * claims 1,2,6 * ---	1-9			
A	EP-A-0 506 386 (SUMITOMO CHEMICAL CO.) * claims 1,2 * ---	1-9			
A	EP-A-0 234 060 (GENERAL ELECTRIC CO.) * claims 1,3,7 * -----	1-9	<table border="1"><tr><td>TECHNICAL FIELDS SEARCHED (Int.Cl.6)</td></tr><tr><td>C08L C08K</td></tr></table>	TECHNICAL FIELDS SEARCHED (Int.Cl.6)	C08L C08K
TECHNICAL FIELDS SEARCHED (Int.Cl.6)					
C08L C08K					
<p>The present search report has been drawn up for all claims</p>					
Place of search	Date of completion of the search	Examiner			
THE HAGUE	14 July 1995	O'Sullivan, T			
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons A : member of the same patent family, corresponding document			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document					